This question paper contains 2 printed pages]

GA-34-2023

FACULTY OF SCIENCE

B.Sc. (Fourth Semester) EXAMINATION APRIL/MAY, 2023

(New Course)

CHEMISTRY

Paper-IX

(Physical and Inorganic Chemistry)

(Thursday, 27-4-2023)

Time: 2.00 p.m. to 4.00 p.m.

Time— Two Hours

Maximum Marks—40

- N.B. := (i) Attempt All questions.
 - (ii) Use of logarithmic table and calculator is allowed.
- 1. Solve any *three* of the following:

15

- (a) What are interhalogen compounds? Explain structure of XY type of interhalogen compound.
- (b) What are oxyacids of halogen? Explain the structure of perhalic acid.
- (c) Explain the structure of Cl₂O and write its uses.
- (d) Define Carbide. Explain its classification.
- (e) Write a short note on Zeolite.
- 2. Solve any *three* of the following:

15

- (a) Derive equation for rate constant of second order chemical reaction for equal concentration of reaction and write unit of rate constant of second order reaction (A=B).
- (b) The half-life period of first order reaction is 15 minutes. How long will it take for completion of 80% of reaction?

P.T.O.

- (c) Explain the Debye-Huckel theory of strong electrolyte.
- (d) State Kohlrausch's law and explain its any two applications.
- (e) State and explain Lambert-Beer's law.
- 3. Solve any two of the following:

10

- (a) Explain any two methods for determination of order of chemical reaction.
- (b) The resistance of 0.02 N solution of salt was found to be 1.5×10^3 ohm. Calculate the specific conductance of solution, if the cell constant is $1.1~{\rm cm}^{-1}$.
- (c) Explain advantages of conductometric titrations.
- (d) What is quantum yield? Describe experimental determination of quantum yield.