This question paper contains 3 printed pages

GA-03-2023

FACULTY OF SCIENCE

B.Sc. (Third Year) (Sixth Semester) EXAMINATION

APRIL/MAY, 2023

(CBCS/Old Pattern)

CHEMISTRY

Paper XIV(A)

(Organic and Inorganic Chemistry)

(Tuesday, 18-04-2023)

Time: 10.00 a.m. to 12.00 noon

Time— Two Hours

Maximum Marks—40

- N.B. := (i) All questions are compulsory.
 - (ii) Figures to the right indicate full marks.
- 1. Answer any *three* of the following:

 $3 \times 5 = 15$

- (a) Write the postulate of valence bond theory.
- (b) Define crystal field spliting. Calculate CFSE of d^1 and d^9 . configuration in strong field octahedral complex.
- (c) Explain the effect of nature of ligand, oxidation state of metal ion on magnitude of crystal field splitting.
- (d) Write different types of electronic transition involved in metal complex.
- (e) Write a note on spectrochemical series.

P.T.O.

2. Answer any three of the following:

 $3\times5=15$

- (a) Interpret IR spectrum of the following compounds
 - (i) Ethane
 - (ii) Ethene
 - (iii) Ethyne.
- (b) Explain non-equivalent proton with example and predict number of PMR signal of:
 - (i) Ethyl amine
 - (ii) Ethyl benzene
 - (iii) di-ethyl ether
- (c) Write the chemical reaction of α -amino acid due to $-NH_2$ group.
- (d) The organic compound having molecular formula C_3H_8O shows the following spectral data :

UV : $\lambda_{\text{max}} = 215 \text{ nm}$

IR: 3600 - 3200 cm⁻¹

 2950 cm^{-1}

 1100 cm^{-1}

PMR (δ ppm):

δ 1.4, d, 6H

 δ 3.5, septet, 1 H

δ 4.6, S, 1H

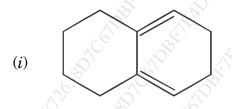
Deduce the structure of compound.

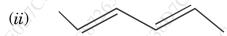
(e) Explain pinacol-pinacolone rearrangement reaction with mechanism.

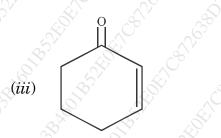
3. Answer any two of the following:

 $2 \times 5 = 10$

(a) Define chromophore and auxochrome. Calculate λ_{max} of :







- (b) Give the synthesis of dipeptide by protecting -COOH group.
- (c) Define chemical shift. Write the advantages of TMS.
- (d) Deduce the structure of compound based on the following PMR spectral data:

 $Molecular\ formula\ :\ C_7H_8$

PMR (d PPM):

δ 2.7, S, 3H

 δ 7.2 – 7.8, m, 5H

GA-03-2023