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WT—09—2024

FACULTY OF SCIENCE

M.Sc. (First Year) (First Semester) EXAMINATION

NOVEMBER/DECEMBER, 2024

(CBCS/New Pattern)

CHEMISTRY

(CH-411)

(Inorganic Chemistry)

(Tuesday, 10-12-2024)

Time : 10.00 a.m. to 1.00 noon

Time—3 Hours

Maximum Marks—75

N.B. :— (i) All questions are compulsory.

(ii) Log table and calculator is allowed.

1. Answer the following (any *three*) : 15

(a) Give the characteristics and examples of associative type mechanism.

(b) Explain ligand substitution reaction and its types.

(c) Define nanomaterial and explain its optical properties

(d) What is charge transfer spectra ? Explain it in metal having mixed valence state.

(e) Calculate the number of microstate for p^1d^1 and 3F .

P.T.O.

2. Answer the following (any *three*) : 15

- (a) Explain in detail electron transfer reaction with suitable example.
- (b) Give an account of inner sphere mechanism.
- (c) Describe in detail scanning tunneling electron microscope.
- (d) Find out ground state term symbol for p^3 .
- (e) Compare Orgel diagram and Tanabe Sugano diagram of d^2 configuration.

3. (a) Describe in detail outer sphere mechanism and mention essential requisite for electron transfer. 8

Or

Explain bottom-up approach for synthesis of nanomaterials.

(b) Explain S_N^1 mechanism with its characteristics. 7

Or

Describe Orgel diagram for d^1 & d^9 configuration.

4. (a) Explain tunneling mechanism in outer sphere mechanism. 8

Or

Give an account on Nephelauxetic effect.

(b) Describe S_{N1} CB mechanism with examples.

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Or

Give the synthesis and classification of nanomaterial.

5. Write short notes on (any *three*) :

15

(a) Bridging ligand in ISM.

(b) DNA and nanomaterial

(c) Bioinorganic nanomaterials

(d) MLCT