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VA—10—2024

FACULTY OF SCIENCE

B.Sc. (Third Year) (Fifth Semester) EXAMINATION

NOVEMBER/DECEMBER, 2024

(CBCS/New Pattern)

CHEMISTRY

Paper—XIII

(Physical and Inorganic Chemistry)

(Monday, 02-12-2024)

Time : 10.00 a.m. to 12.00 noon

Time—2 Hours

Maximum Marks—40

N.B. :- (i) Attempt *all* questions.

(ii) Figures to the right indicate full marks.

(iii) Use of logarithmic table and non-functional calculator is allowed.

1. Answer any *three* of the following : 3×5=15

(a) What are organometallic compounds ? Give their classification with examples.

(b) (i) Write a short note on nomenclature of simple and mixed organometallic compounds with its examples.

(ii) Explain transition metal organometallic compound with its example.

(c) Give the methods of preparation and properties of organolithium compounds.

P.T.O.

- (d) What are metal carbonyls ? Give their classification with examples.
- (c) (i) Draw the structure of $\text{Fe}_3(\text{CO})_{12}$ and $\text{CO}_2(\text{CO})_8$.
- (ii) Describe polynuclear metal carbonyls with examples.

2. Answer any *three* of the following : 3×5=15

- (a) Derive the equation for moment of inertia of rigid diatomic rotor.
- (b) Explain Nernst distribution law when one of solute gets associated.
- (c) What is third order reaction ? Give characteristics of third order reaction.
- (d) The fundamental frequency of CO is 2500 cm^{-1} . Calculate force constant of this molecule. The atomic masses are

$$^{12}\text{C} = 19.0 \times 10^{-27} \text{ kg}$$

$$^{16}\text{O} = 26.0 \times 10^{-27} \text{ kg.}$$

- (e) Explain pure rotational Raman spectra of rigid diatomic molecule.

3. Solve any *two* of the following : 2×5=10

- (a) Explain effect of isotopic substitution on rotational spectra.
- (b) Explain $\sigma - \sigma^*$ and $\pi \rightarrow \pi^*$ transition.
- (c) What are consecutive reactions and opposing reactions ? Explain.

(d) Succinic acid was shaken with mixture of water and ether. After the distribution, upon analysis the concentration of acid in two layers was found as follows :

conc. in aqueous layer 0.0252, 0.071, 0.121

(mole lit⁻¹)

conc. in ether layer 0.0046, 0.013, 0.022.

(mole. lit⁻¹)

Comment on the result. What is molecular state of acid in ether ?