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PA—13—2024

FACULTY OF SCIENCE

B.Sc. (First Year) (First Semester) EXAMINATION

APRIL/MAY, 2024

(New/CBCS Pattern)

CHEMISTRY

Paper-II

(Physical and Inorganic Chemistry)

(Wednesday, 10-04-2024)

Time : 10.00 a.m. to 12.00 noon

Time—2 Hours

Maximum Marks—40

N.B. :— (i) Attempt *all* questions.

(ii) Use of calculator and logarithmic table is allowed.

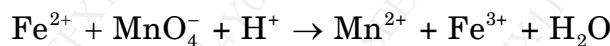
1. Answer any *three* of the following : 15

(a) Explain the basic strength of Hydroxides of alkali and alkaline earth metals.

(b) Write a note on formation of complexes of alkali metal with Salicylaldehyde and Acetyl acetone.

P.T.O.

- (c) Discuss in brief the formation of carbonate and bicarbonate of s-block elements.
- (d) Define oxidation, reduction, oxidizing agent and reducing agent according to oxidation number concept.
- (e) Balance the following equation by Ion-electron method :



2. Answer any *three* of the following :

15

- (a) Prove that, $\text{pH} + \text{pOH} = 14$.

Calculate the pH of 0.025 M. HCl solution.

- (b) Derive an expression for critical constants in terms of van der Waals constants ‘ a ’ and ‘ b ’.
- (c) Discuss the factors affecting adsorption.
- (d) State and explain the law of rational indices and write a note on ‘Miller Indices’.
- (e) Explain the deviation of real gases from ideal behaviour.

3. Answer any two of the following : 10

- (a) State and explain the term ‘Permutation’. Evaluate ${}^{16}\text{P}_2$.
- (b) Explain the determination of crystal structure of Potassium chloride (KCl) by Bragg’s X-ray diffraction method.
- (c) What is adsorption isotherm ? Explain Freundlich adsorption isotherm.
- (d) Calculate the Root Mean Square (RMS) velocity of N_2 and CO_2 molecules at 300 K. (Given : $R = 8.314 \text{ JK}^{-1}\text{mol}^{-1}$)