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**PA—11—2024**

**FACULTY OF SCIENCE**

**B.Sc. (Third Year) (Fifth Semester) EXAMINATION**

**MARCH/APRIL, 2024**

**(CBCS/New Pattern)**

**CHEMISTRY**

**Paper-XIII**

**(Physical and Inorganic Chemistry)**

**(Monday, 08-04-2024)**

**Time : 10.00 a.m. to 12.00 noon**

*Time—2 Hours*

*Maximum Marks—40*

*N.B. :—* (i) Attempt *all* questions.

(ii) Use logarithmic table and non-programmable calculator is allowed.

1. Answer any *three* of the following :

3×5=15

(a) Explain polymerization of  $W^{6+}$ .

(b) Draw the structure of  $CrO_4^{2-}$  tetrahedral heteropoly anion.

(c) Explain the structure of isopoly anions of  $Mo^{6+}$ .

(d) Draw and explain the structure of  $Ir(CO)_{12}$ .

(e) Define isolobal fragments and explain  $P_4$  fragment.

P.T.O.

2. Solve any *three* of the following : 3×5=15

- (a) Derive the Ilkovic equation for the diffusion current in a polarographic cell.
- (b) Describe in brief any *two* applications of Polarography.
- (c) Define magnetic susceptibility, specific susceptibility and give their units.
- (d) State and explain Raoult's law for vapour pressure of binary solutions of volatile liquids.
- (e) Derive Gibbs-Duhem-Margules equation.

3. Answer any *two* of the following : 2×5=10

- (a) Derive an expression for chemical potential for ideal solution.
- (b) Derive an expression for free energy change of mixing of an ideal solution.
- (c) Explain the effect of temperature on paramagnetic, diamagnetic and ferromagnetic substances.
- (d) Explain the principle of Polarography.