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PA—01—2024

FACULTY OF SCIENCE

B.Sc. (Third Year) (Sixth Semester) EXAMINATION

APRIL/MAY, 2024

(New/CBCS Pattern)

CHEMISTRY

Paper XIV(A₁)

(Organic and Inorganic Chemistry)

(Tuesday, 2-4-2024)

Time : 10.00 a.m. to 12.00 noon

Time—Two Hours

Maximum Marks—40

N.B. :— (i) All questions are compulsory.

(ii) Figures to the right indicate full marks.

1. Answer any *three* of the following : 3×5=15
- (a) What are inner orbital complex ? Explain with suitable example.
- (b) Define CFSE. Calculate CFSE of d^4 and d^8 configuration in weak ligand field octahedral complex.
- (c) Explain the effect of size of metal ion geometry of complex on magnitude of crystal field splitting.
- (d) What is d-d transition ? Write its selection rule.
- (e) Discuss electronic spectrum of $[\text{Ti}(\text{H}_2\text{O})_6]^{3+}$ complex.

P.T.O.

2. Answer any *three* of the following :

3×5=15

(a) Interpret IR spectrum of the following compounds :

(i) Ethyne

(ii) *t*-butyl alcohol

(iii) Acetophenone.

(b) Define spin-spin splitting. Predict the number of NMR signal of :

(i) Methanol

(ii) Ethylamine

(iii) Diethyl ether.

(c) Define copolymer. Explain anionic polymerization with mechanism.

(d) Explain photofries rearrangement reaction with mechanism.

(e) The organic compound having molecular formula $C_4H_{10}O$ shows the following spectral data :

UV : Transparent $\lambda_{max} = 210 \text{ nm}$

IR : $3600 - 3200 \text{ cm}^{-1}$

2950 cm^{-1}

1150 cm^{-1}

PMR : (δ ppm)

$\delta 1.5, \text{ s}, 9\text{H}$

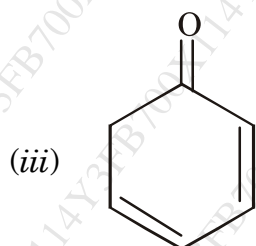
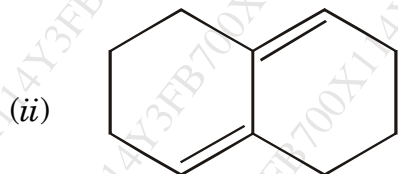
$\delta 4.5, \text{ s}, 1\text{H}$

Deduce the structure of compound.

3. Answer any *two* of the following :

2×5=10

(a) Define bathochromic and hypsochromic effect and calculate λ_{\max} of :



(b) Define chemical shift. Write the advantages of TMS.

(c) Give the synthesis and uses of neoprene.

(d) Deduce the structure of compound based on the following PMR spectral data :

Molecular formula — $C_4H_8O_2$

PMR (δ ppm) :

δ 0.9, t, 3H

δ 2.1, q, 2H

δ 3.9, s, 3H.