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## NEPRT—143—2024

## FACULTY OF SCIENCE

## M.Sc. (NEP) (First Year) (Second Semester) EXAMINATION APRIL/MAY, 2024

**CHEMISTRY** 

Paper-II-SCHEC-453

(Physical Chemistry)

(Tuesday, 23-04-2024)

Time: 10.00 a.m. to 1.00 p.m.

Time—3 Hours

Maximum Marks—80

- N.B. := (1) Q. No. 1 is compulsory.
  - (2) Solve any three questions from Q. Nos. 2 to 6.
  - (3) Use of log table and calculator is allowed.
  - (4) Figures to the right indicate full marks.
- 1. Solve the following:

20

- (a) What are surface active agents? Discuss their classification in detail with examples.
- (b) Define  $\overline{M}_M$  and  $\overline{M}_N$ . Determine  $\overline{M}_N$  and  $\overline{M}_M$  of a polymer sample containing of equal number of molecules with  $M_1=20{,}000$  and  $M_2=2{,}00{,}000$ . Also calculate polydispersity index.
- (c) What is half-wave potential in polarography? Explain its importance.
- (d) Describe Lindemann's theory of unimolecules reactions.

P.T.O.

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2.	Atten	apt the following:						20		
	(a)	Derive Gibbs adsor	ption is	sotherm.	Explain in	detail its	significa	nce.		
	( <i>b</i> )	What is current exchange density? Explain Tafel Plot.								
3.	Solve	: 100 <sub>00</sub> 2411						20		
	(a)	Write an account o	n liqui	d crystal	polymers.	Discuss an	Osmomo	etry		
	method of molecular weight determination of polymers.									
	(b)	What is Differen	tial m	ethod for	r determi	nation of	order c	of a		
		reaction ?								
		At a certain temp	peratur	e the ha	lf-life per	riods for t	he catal	ytic		
		decomposition of ammonia were found to be as:								
		$p_{ m (mm\ of\ Hg)}$	50	-02°T	100	200				
		t <sub>1/2 (min)</sub>	3.8	52	1.92	1.01				
		Find out order, n								
4.	Solve	the following:						20		
	(a)	State BET equation for multilayer adsorption and explain its significance								
		How is it used for the estimation of surface area of an adsorbent.								
	(b)	What is intrinsic viscosity? The intrinsic viscosity of a solution of								
		polyisobutylene at 20°C is 180 cm $^3$ /gm. If [ $\eta$ ] is related to viscosity-								
		average molar mass	s by eq	uation, [η	] = 3.60 ×	$10^{-2} [m]^{0}$	629 calcu	late		

the molar mass of polymer.

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5.	Solve	:							20
	$(\alpha)$	Derive	Butler-Va	olmer ea	uation of	Pelectrode	kinetic re	eactions	

- What is enzyme catalysis? Derive Michaelis-Menten equation and (*b*)
- explain its importance.
- Surface films on liquids (a)

Write short notes on the following:

6.

- Corrosion monitoring and its prevention methods (*b*)
- Kinetics of pyrolysis of acetaldehyde (c)
- Electrically conducting and fire resistant polymers. (*d*)