



UNIT: IV

Quantum Theory of Hydrogen

BY
Bhanudas Narwade
Asst. Prof
Degloor college,
Degloor



The hydrogen atom is the simplest physical system *containing interaction potentials* (i.e., not just an isolated particle).

Simple: one proton, one electron, and the electrostatic (Coulomb) potential that holds them together.

The potential energy in this case is just

$$V = - \frac{e^2}{4\pi\epsilon_0 r}$$

(the attractive potential between charges of $+e$ and $-e$, separated by a distance r).

This is a stationary state potential (no time dependence). We could just plug it in to Schrödinger's equation to get

$$\frac{\partial^2 \psi}{\partial x^2} + \frac{2m}{\hbar^2} \left[E - \left(- \frac{e^2}{4\pi\epsilon_0 r} \right) \right] \psi = 0$$



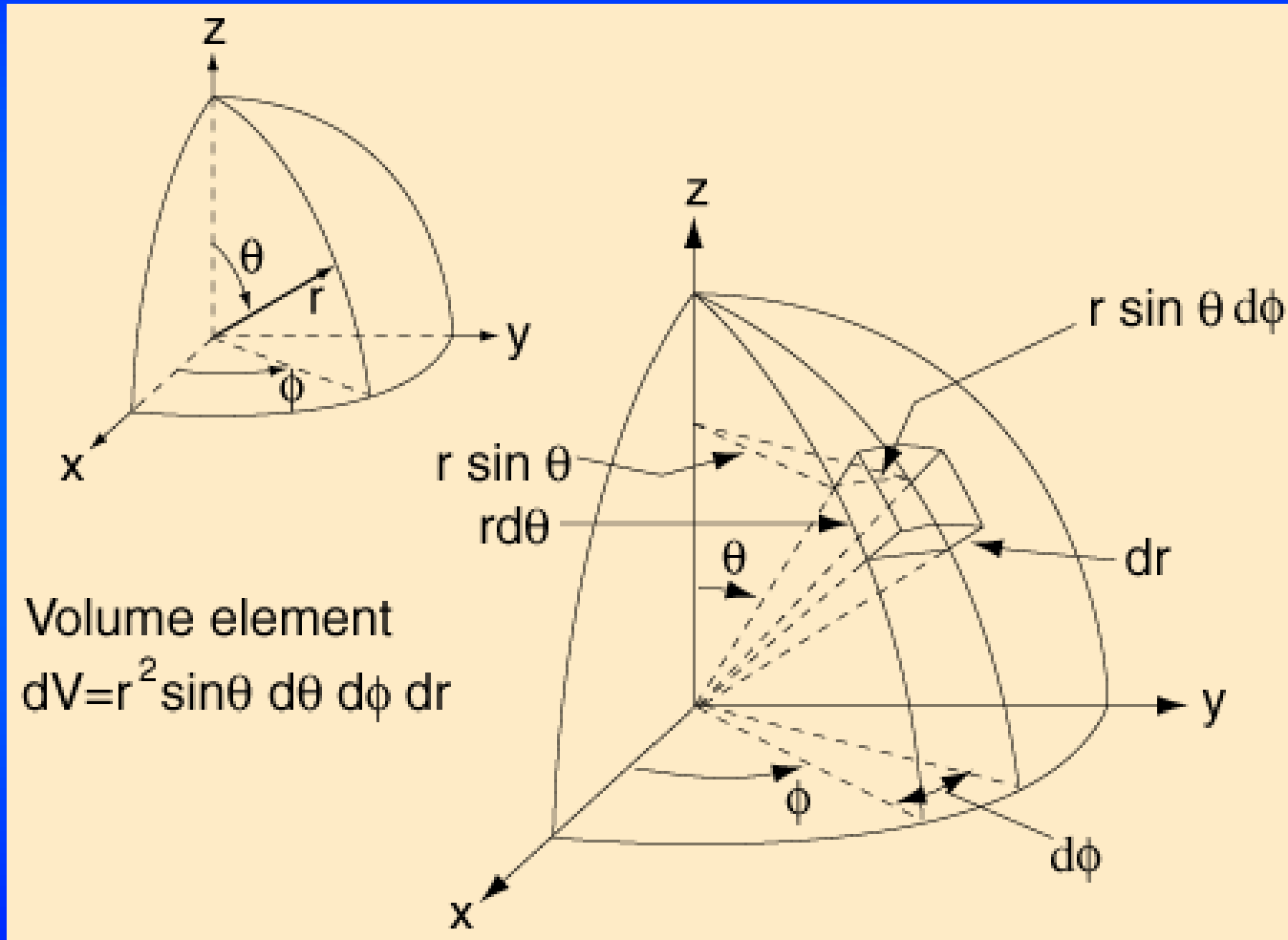
The potential looks quite simple, but it is a function of r , not x or (xyz) . What can we do about that?

$$x^2 + y^2 + z^2 = r^2 \Rightarrow r = \sqrt{(x^2 + y^2 + z^2)}$$

$$\frac{\partial^2 \psi}{\partial x^2} + \frac{2m}{\hbar^2} \left[E - \left(- \frac{e^2}{4\pi\epsilon_0 \sqrt{(x^2 + y^2 + z^2)}} \right) \right] \psi = 0$$



The spherically symmetric potential “tells” us to use spherical polar coordinates!



<http://hyperphysics.phy-astr.gsu.edu/hbase/sphc.html>



In spherical polar coordinates, r is the length of the radius vector from the origin to a point (xyz)

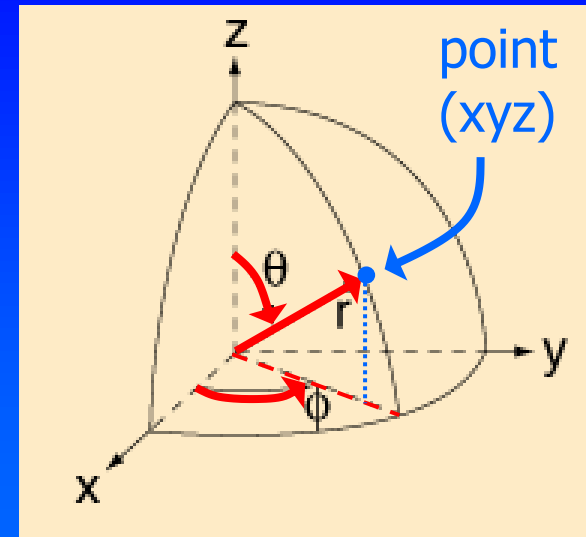
$$r = \sqrt{(x^2 + y^2 + z^2)},$$

θ is the angle between the radius vector and the $+z$ axis

$$\theta = \cos^{-1} \left(\frac{z}{\sqrt{(x^2 + y^2 + z^2)}} \right),$$

and ϕ is the angle between the projection of the radius vector onto the xy plane and the $+x$ axis

$$\phi = \tan^{-1} \left(\frac{y}{x} \right).$$





The equations on the previous slide tell us how to express $(r\theta\phi)$ in terms of (xyz) . We can also express (xyz) in terms of $(r\theta\phi)$:

$$x = r \sin(\theta) \cos(\phi), y = r \sin(\theta) \sin(\phi), z = r \cos(\theta).$$

Now we can re-write the 3D Schrödinger equation

$$\frac{\partial^2 \psi}{\partial x^2} + \frac{\partial^2 \psi}{\partial y^2} + \frac{\partial^2 \psi}{\partial z^2} + \frac{2m}{\hbar^2} (E - V) \psi = 0$$

In three dimensions, and in spherical polar coordinates, as

$$\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial \psi}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \psi}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \psi}{\partial \phi^2} + \frac{2m}{\hbar^2} (E - V) \psi = 0.$$



$$\sin^2\theta \frac{\partial}{\partial r} \left(r^2 \frac{\partial \psi}{\partial r} \right) + \sin\theta \frac{\partial}{\partial \theta} \left(\sin\theta \frac{\partial \psi}{\partial \theta} \right) + \frac{\partial^2 \psi}{\partial \phi^2} + \frac{2mr^2 \sin^2\theta}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) \psi = 0.$$

This equation gives us the wave function ψ for the electron in the hydrogen atom. If we can solve for ψ , in principle we know "everything" there is to know about the hydrogen atom.



Separation of Variables

A differential equation for each variable

Schrodinger's equation in spherical polar coordinates for hydrogen atom can be separated into three independent equations involving each single coordinate only

$\psi(r, \theta, \phi)$ has a form of product of three different functions $R(r)$, $\Theta(\theta)$, $\Phi(\phi)$

Hydrogen Atom Wavefunction: $\psi(r, \theta, \phi) = R(r) \Theta(\theta) \Phi(\phi)$

The function $R(r)$ describes how the wave function ψ of electron varies along radius vector from nucleus with θ ,
And ϕ constant



The function $\Theta(\theta)$ describes how the wave function ψ of electron varies with zenith angle θ along a meridian on a sphere centered at the nucleus with r And ϕ constant

The function $\Phi(\phi)$ describes how the wave function ψ of electron varies with azimuth angle ϕ along a parallel on a sphere centered at the nucleus with r And θ constant

Simply

$$\psi = R \Theta \Phi$$



Thus the partial derivatives in Schrödinger's equation become

$$\frac{\partial \psi}{\partial r} = \Theta \Phi \frac{\partial R}{\partial r} = \Theta \Phi \frac{dR}{dr}$$

$$\frac{\partial \psi}{\partial \theta} = R \Phi \frac{\partial \Theta}{\partial \theta} = R \Phi \frac{d\Theta}{d\theta}$$

$$\frac{\partial^2 \psi}{\partial \phi^2} = R \Theta \frac{\partial^2 \Phi}{\partial \phi^2} = R \Theta \frac{\partial^2 \Phi}{\partial \phi^2}$$

The partial derivatives become full derivatives because R , Θ , and Φ depend on r , θ , and ϕ only.



Schrodinger's equation for H atom

$$\sin^2\theta \frac{\partial}{\partial r} \left(r^2 \frac{\partial \psi}{\partial r} \right) + \sin\theta \frac{\partial}{\partial \theta} \left(\sin\theta \frac{\partial \psi}{\partial \theta} \right) + \frac{\partial^2 \psi}{\partial \phi^2} + \frac{2mr^2 \sin^2\theta}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) \psi = 0.$$

Substituting $\psi = R\Theta\Phi$ into Schrödinger's equation and divide by $R\Theta\Phi$. The result is

$$\frac{\sin^2\theta}{R} \frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) + \frac{\sin\theta}{\Theta} \frac{d}{d\theta} \left(\sin\theta \frac{d\Theta}{d\theta} \right) + \frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2} + \frac{2mr^2 \sin^2\theta}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) = 0.$$



Third term of above eq. is function of azimuth angle
,Rearranging above equation

$$\frac{\sin^2\theta}{R} \frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) + \frac{\sin\theta}{\theta} \frac{d}{d\theta} \left(\sin\theta \frac{d\theta}{d\theta} \right) + \frac{2mr^2 \sin^2\theta}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) = - \frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2}.$$

Differential eq. for function ϕ is

$$\frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2} = m_l^2$$



Dividing above eq. by $\sin^2 \theta$ and rearranging

$$\frac{1}{R} \frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) + \frac{2mr^2}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) = \frac{m_l^2}{\sin^2 \theta} - \frac{1}{\Theta \sin \theta} \frac{d}{d\theta} \left(\sin \theta \frac{d\Theta}{d\theta} \right)$$

The equation for function Θ and R are

$$\frac{m_l^2}{\sin^2 \theta} - \frac{1}{\Theta \sin \theta} \frac{d}{d\theta} \left(\sin \theta \frac{d\Theta}{d\theta} \right) = l(l + 1)$$

$$\frac{1}{R} \frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) + \frac{2mr^2}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) = l(l + 1)$$



Equation for Φ

$$\frac{\partial^2 \Phi}{\partial \phi^2} = m_l^2 \Phi$$

Equation for Θ

$$\frac{1}{\Theta \sin \theta} \frac{d}{d\theta} \left(\sin \theta \frac{d\Theta}{d\theta} \right) + \left[l(l+1) - \frac{m_l^2}{\sin^2 \theta} \right] \Theta = 0$$

Equation for Φ

$$\frac{1}{r^2} \frac{d}{dr} \left(r^2 \frac{dR}{dr} \right) + \left[\frac{2m}{\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0 r} + E \right) - \frac{l(l+1)}{r^2} \right] R = 0$$