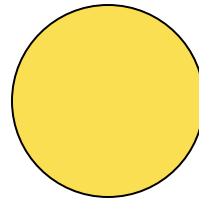


# Phase Velocity and Group Velocity

# Particle

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Our traditional understanding of a particle...

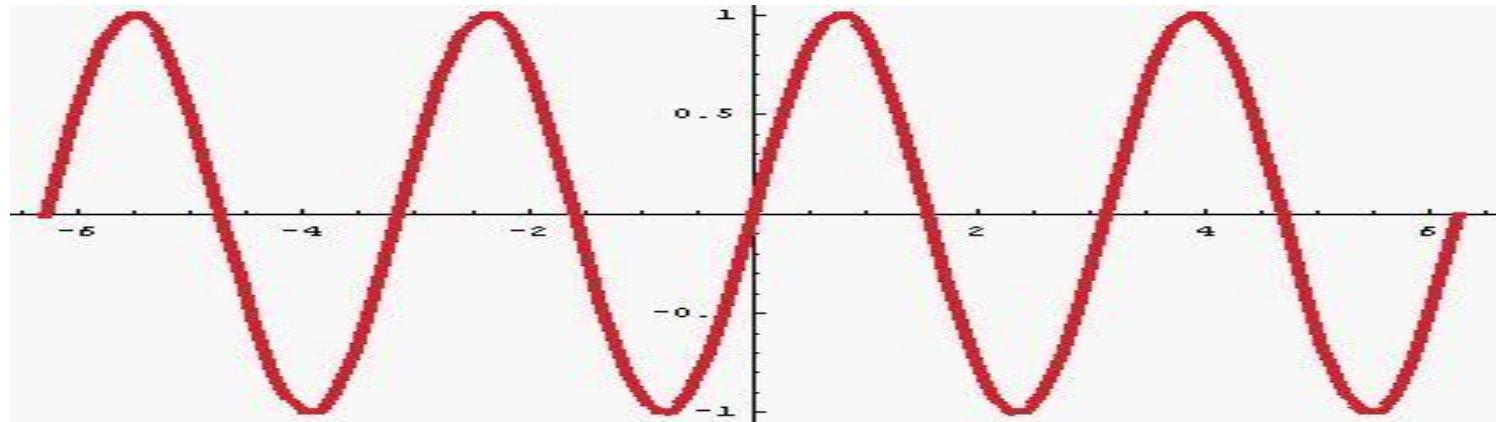


“Localized” - definite position, momentum,  
confined in space

# Wave

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Our traditional understanding of a wave....



“de-localized” – spread out in space and time

# Waves

## Normal Waves

- are a disturbance in space
- carry energy from one place to another
- often (but not always) will (approximately) obey the classical wave equation

## Matter Waves

- disturbance is the wave function  $\Psi(x, y, z, t)$ 
  - probability amplitude  $\Psi$
  - probability density  $p(x, y, z, t) = |\Psi|^2$

# General Wave properties

Oscillations at a particular point

$$y = A \cos 2 \pi \nu t$$

travelling waves  $x = V_p t$

$$y = A \cos 2 \pi \nu (t - x/V_p)$$

$$y = A \cos 2 \pi (\nu t - \nu x/V_p)$$

$$\text{but } V_p = \nu \lambda$$

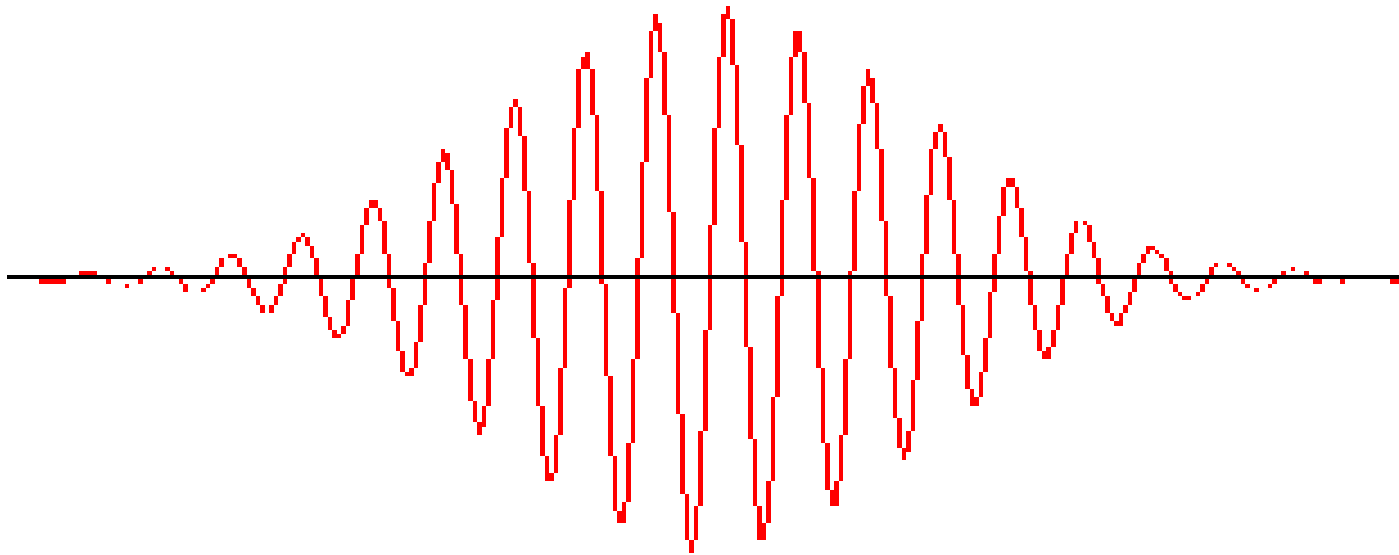
$$y = A \cos(\omega t - kx)$$

$$\omega = 2\pi\nu \text{ (angular frequency)}$$

$$k = \frac{2\pi}{\lambda} \text{ (wave number)}$$

# A “Wave Packet”

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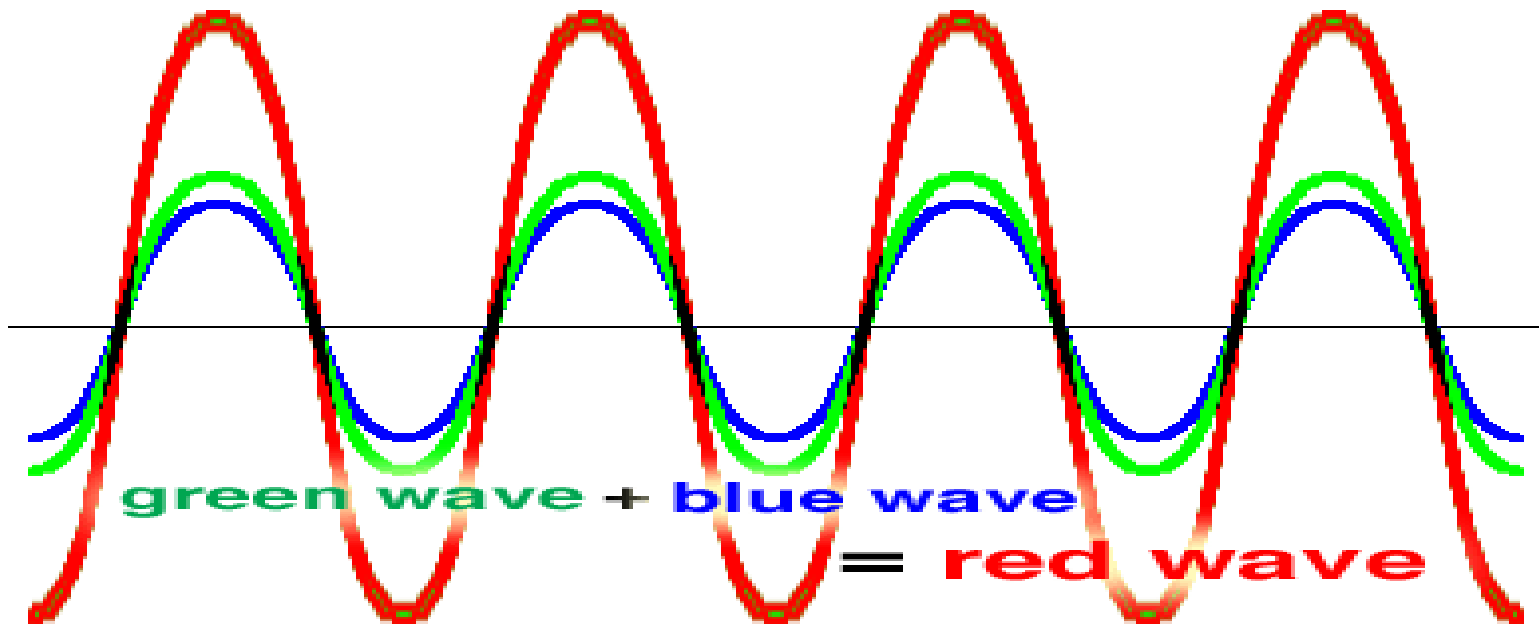


How do you construct a wave packet?

What happens when you add up waves?

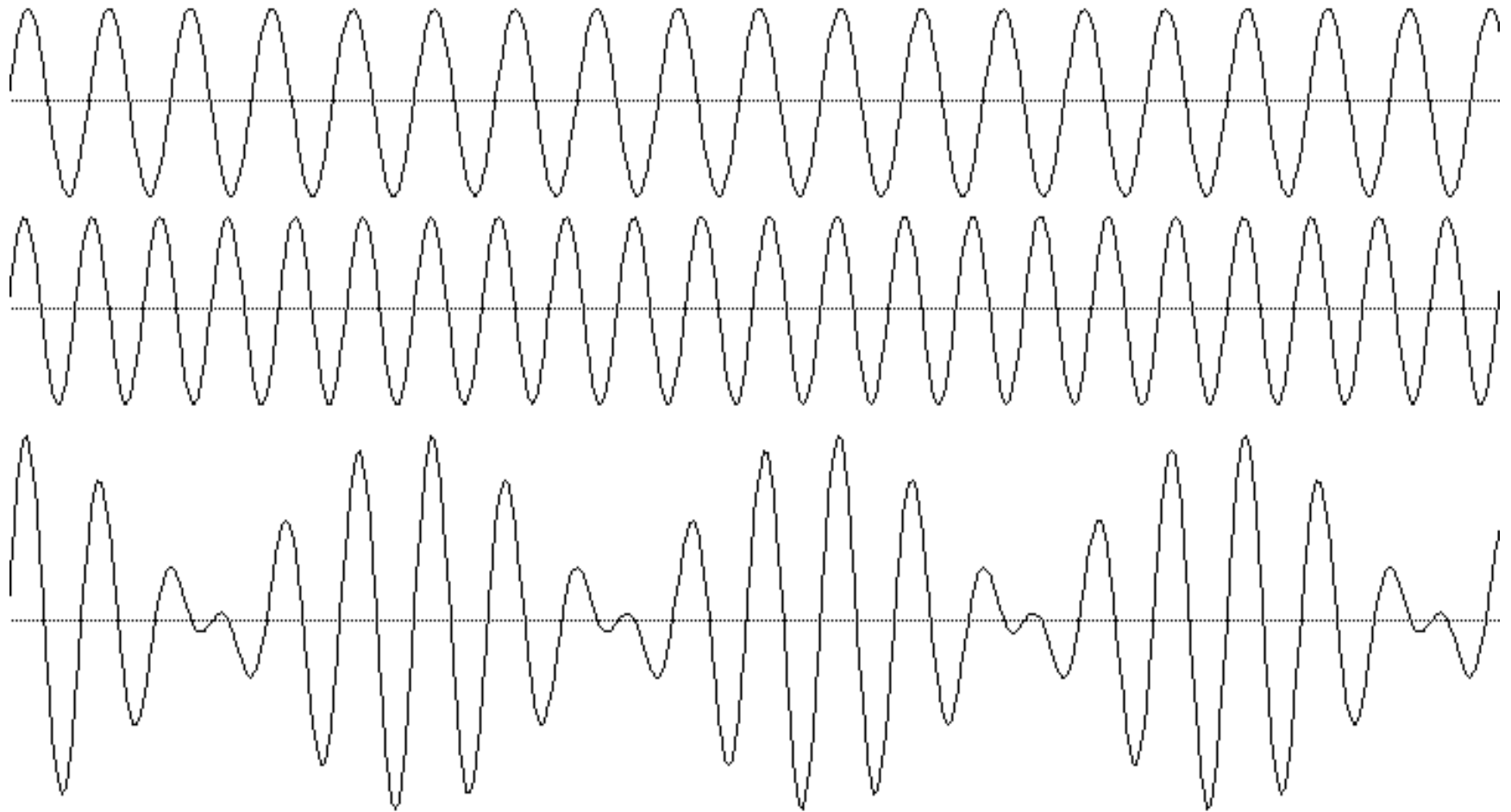
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## The Superposition principle



# Adding up waves of different frequencies.....

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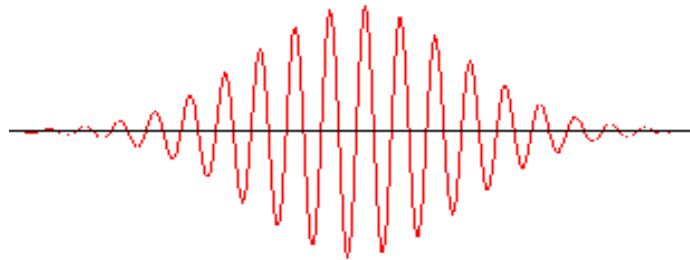




# Constructing a wave packet by adding up several waves .....

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If several waves of different wavelengths (frequencies) and phases are superposed together, one would get a resultant which is a **localized wave packet**



# A wave packet describes a particle

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- A **wave packet** is a group of waves with slightly different wavelengths interfering with one another in a way that the amplitude of the group (envelope) is non-zero only in the neighbourhood of the particle
- A wave packet is **localized** – a good representation for a particle!

# Wave packet, phase velocity and group velocity

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- The spread of **wave packet** in wavelength depends on the required **degree of localization** in space – the central wavelength is given by

$$\lambda = \frac{h}{p}$$

- What is the **velocity** of the wave packet?

# Wave packet, phase velocity and group velocity

- The velocities of the individual waves which superpose to produce the wave packet representing the particle are different - the **wave packet as a whole** has a different velocity from the waves that comprise it
- **Phase velocity**: The rate at which the phase of the wave propagates in space
- **Group velocity**: The rate at which the envelope of the wave packet propagates

## Phase velocity

- Phase velocity is the rate at which the phase of the wave propagates in space.
- This is the velocity at which the phase of any one frequency component of the wave will propagate.
- You could pick one particular phase of the wave and it would appear to travel at the phase velocity.
- The phase velocity is given in terms of the wave's angular frequency  $\omega$  and wave vector  $k$  by

$$v_P = \frac{\omega}{k}$$

# Group velocity

Group velocity of a wave is the velocity with which the variations in the shape of the wave's amplitude (known as the **modulation** or **envelope** of the wave) propagate through space.

The group velocity is defined by the equation

$$v_g \equiv \frac{\partial \omega}{\partial k},$$

where:

$v_g$  is the group velocity;

$\omega$  is the wave's angular frequency;

$k$  is the wave number.

The function  $\omega(k)$ , which gives  $\omega$  as a function of  $k$ , is known as the dispersion relation.

# Velocity of De-Broglie Waves-Phase velocity

- Moving material particle-De broglie said a wave is associated which travel with the same velocity as that of the wave.
- Let De-broglie wave velocity is  $V_p$ , so

$$V_p = v\lambda$$

# Phase Velocity - continued

- we know that

$$\lambda = h/mv$$

- From Planck's Law

$$E = h\nu$$

- From mass energy relation

$$E = mc^2$$



# Phase Velocity - continued

- $h\nu = mc^2$        $v = mc^2/h$

$$V_p = v \lambda$$

Therefore

$$V_p = mc^2/h \times h/mv$$

- $V_p = C^2/v$
- So Particle velocity must be less than the velocity of light.  $V_p > c$
- We can say that De-Broglie waves must travel faster than the velocity of light.

# Group Velocity=Particle velocity

$$y_1 = A \cos(\omega t - kx)$$

$$y_2 = A \cos[(\omega + d\omega)t - (k + dk)x]$$

$$y = y_1 + y_2$$

$$y = A \cos(\omega t - kx) + A \cos[(\omega + d\omega)t - (k + dk)x]$$

$$y = 2A \cos \left[ \frac{(2\omega + d\omega)t}{2} + \frac{(2k + dk)x}{2} \right] \cos \left[ \frac{(d\omega)t}{2} - \frac{(dk)x}{2} \right]$$

with  $d\omega \ll \omega, dk \ll k$

$$y \cong 2A \cos \left[ \frac{d\omega}{2} t - \frac{dk}{2} x \right] \cos[\omega t - kx] \text{----- (1)}$$

Equation 1 represent a wave of angular velocity  $\omega$  and wave number  $k$  which has superimposed upon it a wave (the process is called modulation) of angular velocity  $\frac{d\omega}{2}$  and wave number  $\frac{dk}{2}$

phase velocity= wave velocity of carrier :  $v_p = \frac{\omega}{k}$

group velocity= wave velocity of envelope :  $v_g = \frac{\Delta\omega}{\Delta k}$

for more than two wave contributions:  $v_g = \frac{d\omega}{dk}$

# De Broglie waves for massive particles

$$\omega = 2\pi\nu$$

$$\omega = 2\pi \frac{E}{h}$$

$$\omega = 2\pi \frac{mc^2}{h}$$

$$\omega = 2\pi \frac{m_0 c^2}{h \sqrt{1 - \frac{v^2}{c^2}}}$$

$$v_g = \frac{d\omega}{dk} = \frac{d\omega/dv}{dk/dv}$$

$$\omega = \frac{2\pi}{h} \frac{m_0 c^2}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$$d\omega/dv = \frac{2\pi m_0 c^2}{h} \left(-\frac{1}{2}\right) \left(1 - \frac{v^2}{c^2}\right)^{-3/2} \left[-2v/c^2\right]$$

$$d\omega/dv = \frac{2\pi}{h} \frac{m_0 v}{\left(1 - \frac{v^2}{c^2}\right)^{3/2}}$$

$$k = \frac{2\pi}{h} \frac{m_0 v}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$$\frac{dk}{dv} = \frac{2\pi m_0}{h} \left[ \frac{1}{\sqrt{1 - \frac{v^2}{c^2}}} + \left(-\frac{1}{2}\right) \left(1 - \frac{v^2}{c^2}\right)^{-3/2} \left[-2v/c^2\right] \right]$$

$$\frac{dk}{dv} = \frac{2\pi m_0}{h} \left(1 - \frac{v^2}{c^2}\right)^{-3/2} \left[1 - \frac{v^2}{c^2} + \frac{v^2}{c^2}\right]$$

$$\frac{dk}{dv} = \frac{2\pi m_0}{h} \frac{1}{\left(1 - \frac{v^2}{c^2}\right)^{3/2}}$$

$$v_g = \frac{d\omega}{dk} = \frac{d\omega/dv}{dk/dv}$$

$$d\omega/dv = \frac{2\pi}{h} \frac{m_0 v}{\left(1 - \frac{v^2}{c^2}\right)^{3/2}}$$

$$dk/dv = \frac{2\pi m_0}{h} \frac{1}{\left(1 - \frac{v^2}{c^2}\right)^{3/2}}$$

$$\Rightarrow v_g = v$$

# Scheme

